

# **MOISTURE ADSORPTION ISOTHERMS AND ADSORPTION ISOSTERIC HEAT OF DRY GROUND MEAT**

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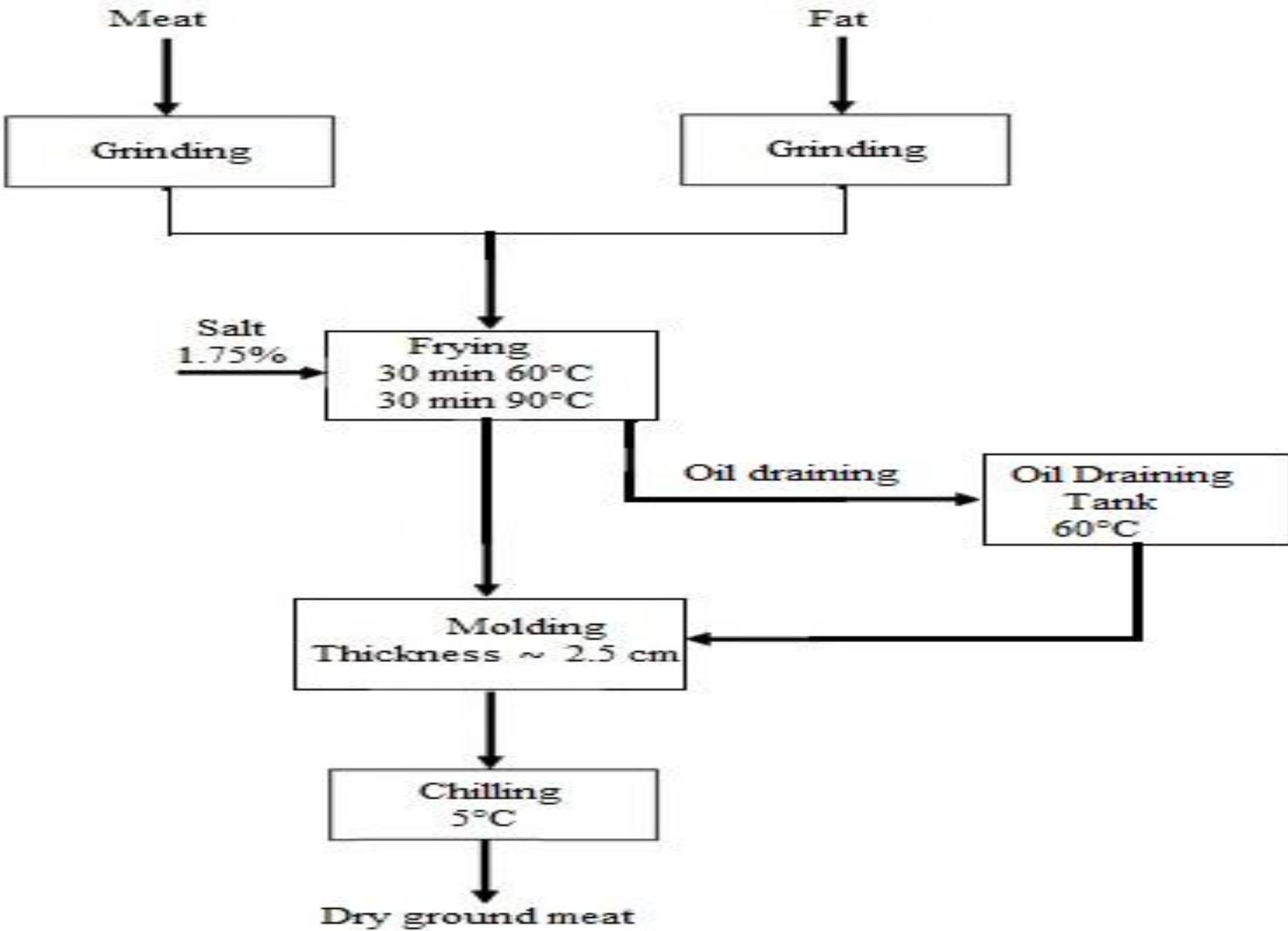


Figure 1. Block diagram of dry ground meat production

## **Materials and Methods**

### **Materials**

Dry ground meat samples with triplicate, which were produced on separate processing days by the same production technique, were purchased from a local company (Kavdirlar, Nevşehir, Turkey).

# Methods

## Physicochemical Analysis

pH

Moisture

Fat

Protein

Ash



(Ockerman, 1985)

Thiobarbituric acid reactive substances (TBARS)

(Lemon, 1975)

# Adsorption isotherms

Saturated salt solutions and relative humidity values

LiCl 11.30%,

KCH<sub>3</sub>COO 23.11%,

MgCl<sub>2</sub> 33.07%,

K<sub>2</sub>CO<sub>3</sub> 43.16%,

Mg (NO<sub>3</sub>)<sub>2</sub> 54.38%,

NaNO<sub>2</sub> 65.40%,

NaCl 75.47%,

KCl 85.11% and

BaCl<sub>2</sub> 90.69%

The temperatures used for the adsorption isotherms:

5, 15 and 25°C.

## Fitting of adsorption data to various isotherm equations

The experimental data obtained were applied to different isotherm equations that are as follows: Iglesias-Chirife, Oswin, BET, Harkins-Jura, Smith, Henderson, Halsey, GAB, Peleg, modified Chung-Pfost, modified Oswin. The goodness of fit as applied to the experimental data were evaluated through the determination coefficient ( $R^2$ ) and mean relative percentage deviation (P).

$$R^2 = 1 - \frac{\text{Residual sum of squares}}{\text{Corrected sum of squares}} \quad (1)$$

$$P = \frac{100}{n} \sum_{i=1}^n \frac{|X_i - X_{pi}|}{X_i} \quad (2)$$

## Isosteric heat of adsorption

The isosteric heats of adsorption were calculated by using Eq. (3), which originates from the Clausius–Clapeyron equation, to the adsorption isotherms at different temperatures.

$$\frac{d \ln a_w}{d \left[ \frac{1}{T} \right]} = - \frac{Q_n^{st}}{R} \quad (3)$$

The isosteric heat of sorption ( $Q^{st}$ ) is calculated by including the latent heat of vaporization for pure water ( $L_r$ ) to the net isosteric heat of sorption (Eq. (4)), considering  $L_r$  as the average value for the temperatures taken into account (2466.18 J/g for 5–25 °C).

$$Q^{st} = Q_n^{st} + L_r \quad (4)$$

# Results and Discussion

## Physicochemical analysis

**Table 1.** Physicochemical characteristics of dry ground meat samples

pH	6.26±0.03
Moisture (%)	51.82±0.10
Protein (%)	24.62±0.18
Fat (%)	21.52±0.48
Ash (%)	2.04±0.22
TBARS ( $\mu\text{mol MDA/ kg}$ )	6.28±0.17

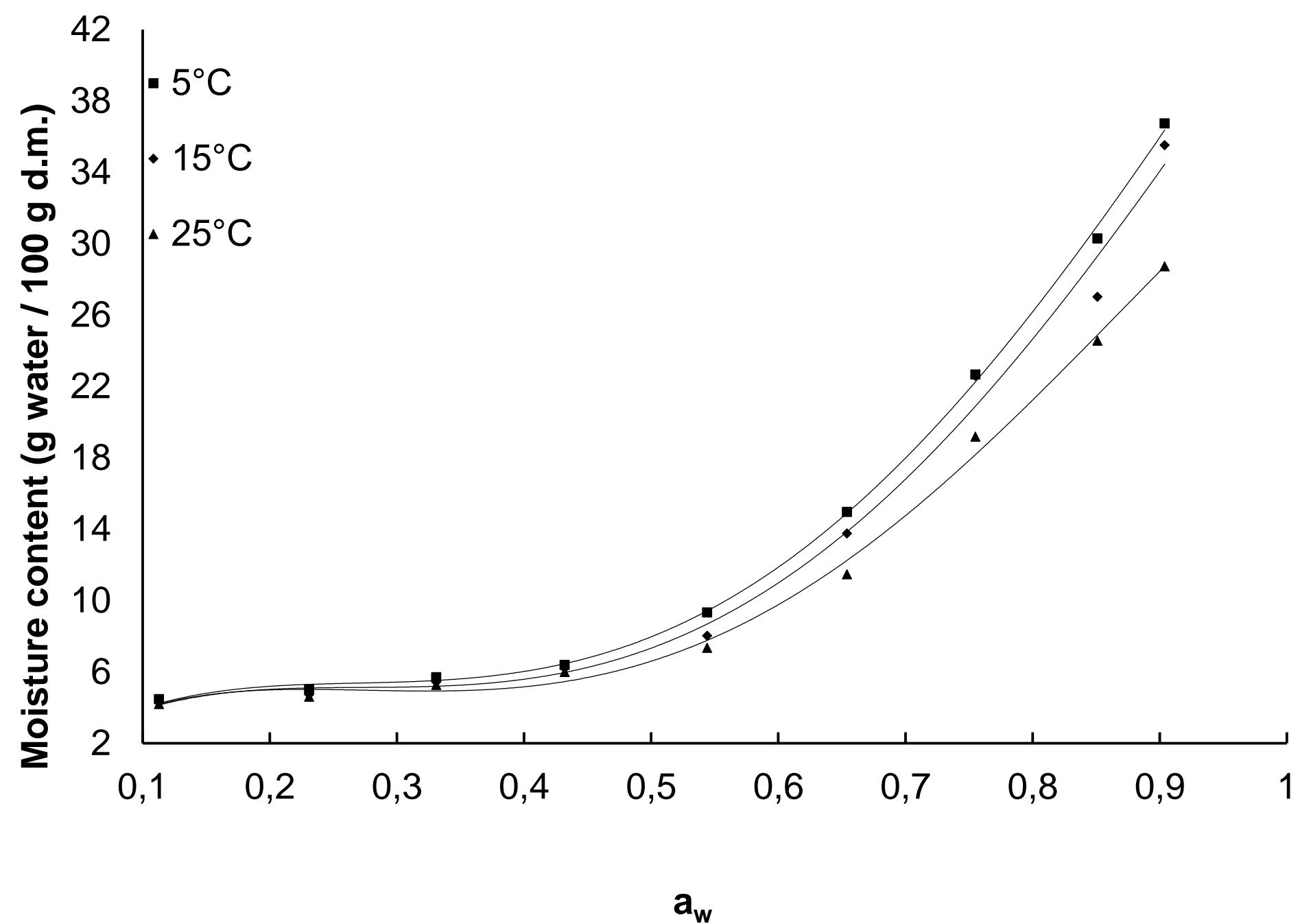
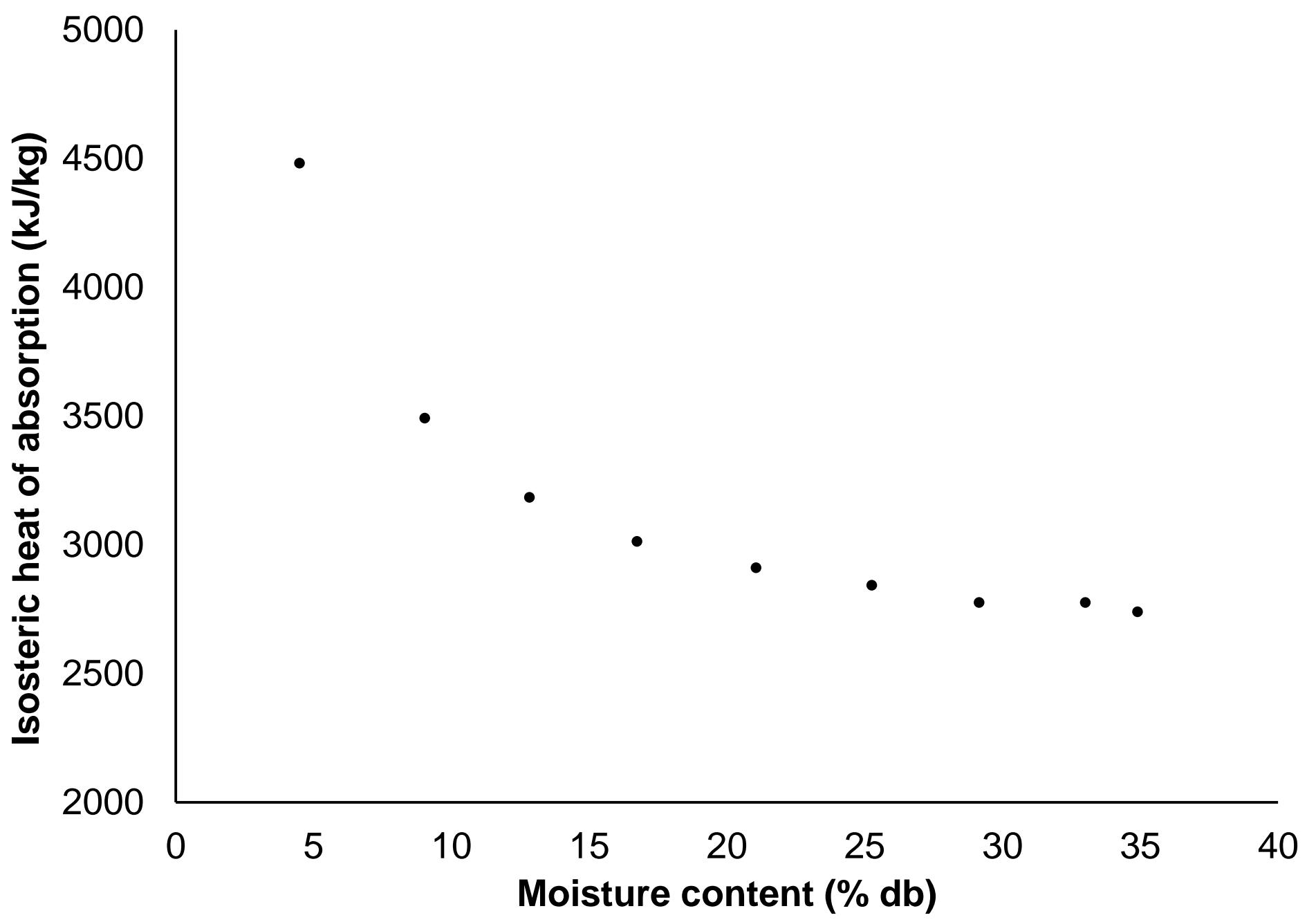


Figure 2. Moisture adsorption isotherms of dry ground meat samples at 5°C, 15°C and 25°C.

Isotherm equations	$a_w$ (range)	Temperature (°C)	Constant parameters	Determination coefficient ( $R^2$ )	P(%)
Iglesias and Chirife <sup>¶</sup> $\ln(m + (m^2 + m_{0.8})^{1/2}) = p + b(a_w)$	0.1-0.9≈	5°C 15°C 25°C≈	b=2.83¶ p=1.60¶ b=2.77¶ p=1.58¶ b=2.57¶ p=1.60≈	0.951¶ 0.938¶ 0.939≈	0.16¶ 0.43¶ 0.29≈
Oswin <sup>¶</sup> $\ln m = \ln k + n \ln\left(\frac{a_w}{1-a_w}\right)$	0.1-0.9≈	5°C 15°C 25°C≈	n=0.56¶ k=10.12¶ n=0.55¶ k=9.55¶ n=0.51¶ k=8.72≈	0.939¶ 0.926¶ 0.929≈	15.44¶ 17.00¶ 15.78≈
BET <sup>¶</sup> $\frac{a_w}{m(1-a_w)} = \frac{1}{(m_0 c)} + \frac{(c-1)}{(m_0 c)} a_w$	0.1-0.9-/0.1-0.5≈	5°C 15°C 25°C≈	C=46.94/60.34¶ m <sub>0</sub> =4.35/4.04¶ C=44.62/166.41¶ m <sub>0</sub> =4.07/3.53¶ C=35.44/-53.75¶ m <sub>0</sub> =3.24/3.32≈	0.866/0.973¶ 0.868/0.998¶ 0.869/0.999≈	12.28/5.76¶ 11.78/5.29¶ 12.33/1.52≈
Harkins-Jura <sup>¶</sup> $\frac{1}{m^2} = \left(\frac{B}{A}\right) - \left(\frac{1}{A}\right) \log a_w$	0.1-0.9≈	5°C 15°C 25°C≈	A=-16.34¶ B=-0.05¶ A=-15.65¶ B=-0.03¶ A=-14.53¶ B=-0.02≈	0.960¶ 0.941¶ 0.960≈	13.37¶ 14.44¶ 13.00≈
Smith <sup>¶</sup> $m = W_b - W \ln(1 - a_w)$	0.1-0.9≈	5°C 15°C 25°C≈	W=-15.60¶ W <sub>0</sub> =0.28¶ W=-14.74¶ W <sub>0</sub> =0.31¶ W=-12.08¶ W <sub>0</sub> =0.49≈	0.978¶ 0.968¶ 0.969≈	19.11¶ 21.22¶ 18.00≈
Henderson <sup>¶</sup> $\ln[-\ln(1 - a_w)] = \ln k + n \ln m$ ≈	0.1-0.9≈	5°C 15°C 25°C≈	n=0.78¶ k=-14.96¶ n=0.77¶ k=-14.03¶ n=0.71¶ k=-12.47≈	0.880¶ 0.861¶ 0.864≈	23.33¶ 26.11¶ 23.55≈
Halsey <sup>¶</sup> $\ln m = \left[\frac{1}{n} \ln c\right] - \left(\frac{1}{n}\right) \ln \left[\ln \frac{1}{a_w}\right]$	0.1-0.9≈	5°C 15°C 25°C≈	n=1.28¶ c=-11.87¶ n=1.30¶ c=-11.45¶ n=1.40¶ c=-12.65≈	0.969¶ 0.961¶ 0.964≈	12.11¶ 12.44¶ 11.22≈

Isotherm equation	$a_w$ (range)	Temperature (°C)	Constant parameters	Determination coefficient ( $R^2$ )	P(%)
GAB	0.1-0.9≈	5°C	$m_m=4.33\pm$ $c=1.01\pm$ $K'=28.23\pm$	0.850±	12.60%
$\frac{a_w}{m} = \alpha a_w^2 + \beta a_w + \gamma$		15°C	$m_m=3.94\pm$ $c=1.03\pm$ $K'=41.07\pm$	0.856±	14.58%
$C = \left[ \beta - \frac{\left(\frac{1}{m_m}\right)}{(-2\gamma)} \right]$		25°C≈	$m_m=3.69\pm$ $c=0.95\pm$ $K'=92.02\pm$	0.891≈	7.13≈
$K = \frac{1}{m_m C \gamma}$					
Peleg	0.1-0.9≈	5°C	A=-46.33±	0.998±	4.49%
$M = A a_w^B + C a_w^D$		15°C	B=-3.37± C=-3.72± D=-0.10±	0.990±	7.25%
		25°C≈	A=-44.36± B=-3.43± C=-3.38± D=-0.14±	0.993≈	6.19≈
Modified Chung-Pfost	0.1-0.9≈	5°C	$a=78.76\pm$ $b=34.24\pm$ $c=0.09\pm$	0.931±	28.22%
$M = -\frac{1}{c} \ln \left[ \frac{(t+b) \ln(a_w)}{-a} \right]$		15°C	$a=34.87\pm$ $b=2.40\pm$ $c=0.92\pm$	0.918±	31.67%
		25°C≈	$a=557.54\pm$ $b=230.96\pm$ $c=0.11\pm$	0.921≈	20.11≈
Modified Oswin	0.1-0.9≈	5°C	$A=-3.86 \times 10^{-4}\pm$ $B=-7.73 \times 10^{-3}\pm$ $C=0.59\pm$	0.978±	14.22%
$\left(\frac{a_w}{1-a_w}\right)^c$		15°C	$A=-7.74 \times 10^{-4}\pm$ $B=-5.16 \times 10^{-3}\pm$ $C=0.60\pm$	0.971±	15.01%
		25°C≈	$A=-1.73 \times 10^{-4}\pm$ $B=-690.73\pm$ $C=0.56\pm$	0.966≈	14.89≈

	Temperature (°C)	BET	GAB
<b>Monolayer moisture content</b>	5	4.35/4.04	4.33
	15	4.07/3.53	3.94
	25	3.24/3.32	3.69



**Figure 3.** Isosteric heat of adsorption of dry ground meat samples

## Conclusion

The sorption isotherms of dry ground meat samples were type II isotherm pattern that is commonly observed in high protein content food products. The equilibrium moisture content relationship of the dry ground meat samples in the studied temperatures were best described by the Iglesias-Chirife and Peleg models.



**THANKS FOR YOUR ATTENTION**

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